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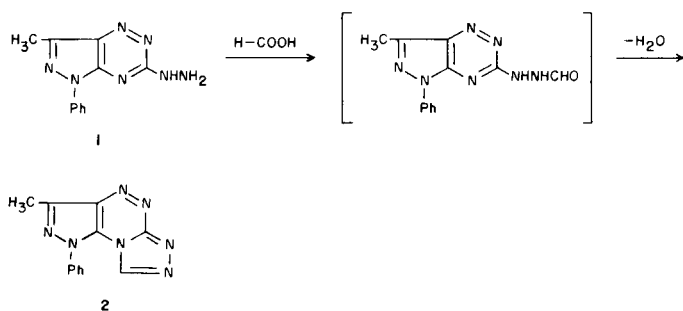
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3-Hydrazino-7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*c*]-as-triazine **1** underwent ring closure and/or condensation reaction with formic acid, acetic acid, acetic anhydride and benzoyl chloride to afford 1*H*-pyrazolo[3,4-*d*]-s-triazolo[3,4-*c*]-as-triazines **2**, **5** and **7a** and/or *N*-acyl derivatives **3**, **4** and **6**. *N*-Acyl derivatives **3** and **6** underwent cyclisation reaction on treatment with phosphoryl chloride to give **5** and **7a**. 3-Methyl-1-phenyl-8-aryl-1*H*-pyrazolo[3,4-*e*]-s-triazolo[3,4-*c*]-as-triazines **7** were also prepared by the reaction of the hydrazono derivatives **8** with thionyl chloride. On treatment of **1** with nitrous acid gave the 8*H*-pyrazolo[3,4-*e*]tetrazolo[5,1-*c*]-as-triazine **9**. Compound **1** underwent ring closure with carbon disulphide or ethyl chloroformate to 1,7-dihydro-8*H*-pyrazolo[3,4-*e*]-s-triazolo[3,4-*c*]-as-triazine derivatives **10** and **12**. Reaction of **1** with ethyl acetoacetate or acetylacetone gave 3-pyrazolo derivatives **13** and **14**.

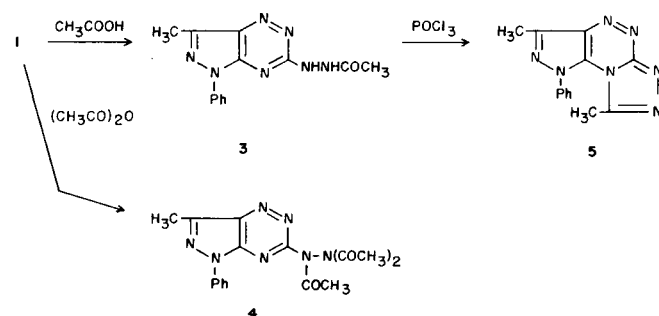
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The broad utility of heterocyclic hydrazines as starting materials for the preparation of several condensed systems containing triazole and tetrazole nuclei has received increasing attention [2-10]. From this view it was of interest to examine the chemistry of 3-hydrazino-7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazine (**1**) [11].

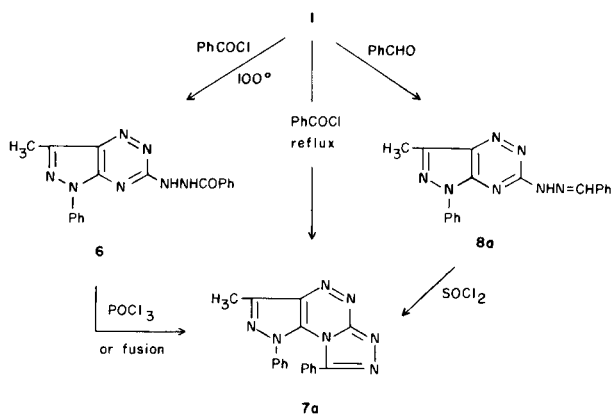
3-Hydrazino-7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazine (**1**) readily underwent ring closure with formic acid to give 3-methyl-1-phenyl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazine (**2**). The reaction proceeds through *N*<sup>2</sup>-acylation followed by thermal cyclisation at N-4 of the triazine ring to form the angular structure **2** as cited in the literature [2,3,12,13]. The chemical structure of **2** was confirmed by analytical and spectroscopic data.



Attempts to synthesise 3,8-dimethyl-1-phenyl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazine (**5**) by refluxing **1** with acetic acid or acetic anhydride were unsuccessful and instead of the target compound **5**, 2-acetyl- and 1,2,2-triacetyl-1-(7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazines **3**, **4** were obtained. The monoacetyl derivative **3** underwent ring closure with phosphoryl chloride [4] in boiling xylene to **5**. The chemical structure of **3**, **4** and **5** was deduced from their analytical and spectroscopic data.

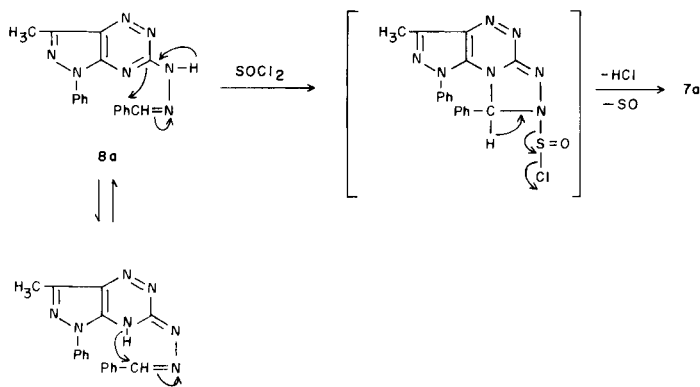


On heating **1** with excess benzoyl chloride at 100° gave 2-benzoyl-1-(7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazine (**6**) was obtained which subsequently cyclised to 3-methyl-1,8-diphenyl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazine (**7a**) by using phosphoryl chloride [5] in boiling xylene or by fusion. Furthermore, the cyclised product **7a** was also obtained directly by refluxing the hydrazino derivative **1** with excess benzoyl chloride [2]. The chemical structure of **6** and **7a** was confirmed by analytical and spectroscopic data.

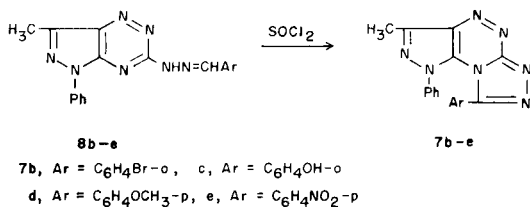


An alternative route for the synthesis of **7a** involved the reaction of benzaldehyde(7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazone (**8a**) [11] with thionyl chloride at reflux for 3 hours. It was found that, the product obtained by this method was identical in all aspects (mp, mmp, ir and pmr) with those obtained by the above methods.

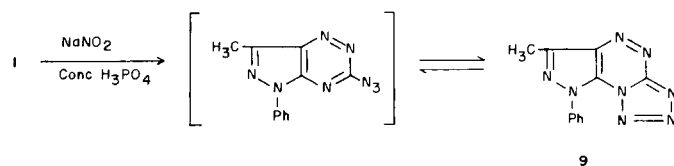
The reaction of **8a** with thionyl chloride presumably proceeds through the following series of transformations.



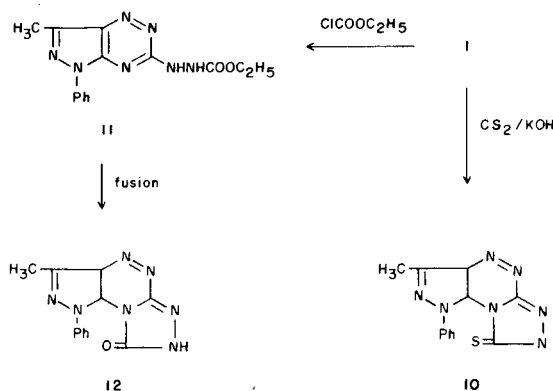
The latter procedure was applied to synthesise 3-methyl-1-phenyl-8-aryl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*]-[1,2,4]triazines **7b-e**. The chemical structure of **7b-e** was confirmed on the basis of analytical data and ir spectra.



Treatment of **1** with nitrous acid gave, in nearly quantitative yield, the corresponding 6-methyl-8-phenyl-8*H*-pyrazolo[3,4-*e*]tetrazolo[5,1-*c*][1,2,4]triazine (**9**). The chemical structure of **9** was confirmed by its elemental analysis and spectroscopic data.



1,7-Dihydro-3-methyl-1-phenyl-8*H*-pyrazolo[3,4-*e*][1,2,4]-triazine-8-thione (**10**) was synthesised by refluxing **1** with carbon disulphide in alcoholic potassium hydroxide [4,5]. On the other hand, refluxing **1** with ethyl chloroformate in pyridine at 100°; alcoholic sodium hydroxide; sodium ethoxide or in benzene and triethylamine produced 2-ethoxycarbonyl-1-(7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazine (**11**). Fusion of **11** above its melting point for 10 minutes eliminates a molecule of alcohol giving 1,7-dihydro-3-methyl-1-phenyl-8*H*-pyrazolo[3,4-*e*]-[1,2,4]triazolo[3,4-*c*][1,2,4]triazin-8-one (**12**). The chemical structures of **10**, **11** and **12** were established on the basis of elemental analyses and spectroscopic data.



The cyclic structure of compounds **2**, **5**, **7**, **9**, **10** and **12** comprised a longer wavelength band in the range (444-560 nm). This band does not appear in the spectra of the starting compound or the intermediates and corresponds to CT transition within the highly conjugated structure formed as a result of cyclization. The CT nature of this band is

Table I

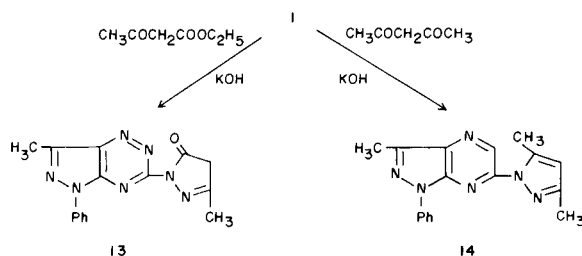
3-Methyl-8-aryl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazines **7b-e**

Compound No.	Mp (°C)	Yield (%)	Solvent of Crystallization [a]	λ max (Dioxane) (ε) nm	Molecular formula	Analysis (%)		
						Calcd./	(Found)	
						C	H	N
<b>7b</b>	238-240	60	B	—	C <sub>18</sub> H <sub>12</sub> N <sub>7</sub> Br [b]	53.22 (53.13)	2.98 (2.96)	24.13 (24.31)
<b>7c</b>	231-233	34	B	—	C <sub>18</sub> H <sub>13</sub> N <sub>7</sub> O	62.97 (63.05)	3.82 (4.00)	28.56 (28.38)
<b>7d</b>	250-252	70	B	286, 522 (46000), (1550)	C <sub>19</sub> H <sub>15</sub> N <sub>7</sub> O	63.86 (63.75)	4.23 (4.31)	27.44 (27.64)
<b>7e</b>	203-205	94	P	286, 344, 510 (35500), (19000), (1750)	C <sub>18</sub> H <sub>12</sub> N <sub>8</sub> O	58.06 (58.19)	3.25 (3.29)	30.09 (30.07)

[a] B = Benzene, P = Pyridine. [b] Calcd.: Br, 19.76%. Found: Br, 19.70%.

evidenced from its broadness as well as its sensitivity towards substitution (*cf.* Experimental).

Reaction of **1** with ethyl acetoacetate and/or acetylacetone in alcoholic potassium hydroxide led to the formation of 7-methyl-3-(3-methyl-2-pyrazolin-1-yl-5-one)-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazine and/or 7-methyl-3-(3,5-dimethylpyrazol-1-yl)-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazine (**13** and **14**) respectively. The chemical structures of **13** and **14** were confirmed by elemental analyses and ir spectra.



## EXPERIMENTAL

All melting points are uncorrected. The ir spectra were recorded on a Perkin-Elmer 599 B spectrophotometer using the potassium bromide wafer technique. The pmr spectra were obtained on a Varian EM-360 (60 MHz) spectrometer. The uv spectra were recorded on a Varian-Cary 219 spectrophotometer.

### 3-Methyl-1-phenyl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazine (**2**).

3-Hydrazino-7-methyl-5-phenyl-5*H*-1,2,4-triazine (**1**) (0.3 g) was refluxed with formic acid (10 ml) for 5 hours. On cooling the reaction mixture and dilution with water, an orange precipitate was formed. The product was crystallised from benzene to give **2** as orange needles on 96% yield, mp 246-248°; ir (potassium bromide): showed no characteristic bands for NH and NH<sub>2</sub> groups; pmr (trifluoroacetic acid): δ 2.2 (s, CH<sub>3</sub>, 3H), 6.5-7.3 (m, aromatic, 5H), 8.4 (s, CH=N, 1H); uv (dioxane): λ max 266 nm (ε 75000), 482 (2566).

*Anal.* Calcd. for C<sub>12</sub>H<sub>9</sub>N<sub>7</sub>: C, 57.37; H, 3.61; N, 39.02. Found: C, 57.74; H, 3.81; N, 39.22.

### 2-Acetyl-1-(7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazine (**3**).

A mixture of **1** (0.3 g) and acetic acid (10 ml) was refluxed for 5 hours. On cooling the reaction mixture a yellow precipitate was formed. The product was crystallised from ethanol to give **3** as pale yellow needles in 79% yield, mp 250-252°; ir (potassium bromide): 3270, 3220 cm<sup>-1</sup> (NH), 1670 (C=O); pmr (DMSO-*d*<sub>6</sub>): δ 1.8 (s, NCOCH<sub>3</sub>, 3H), 2.4 (s, CH<sub>3</sub>, 3H), 6.8-7.7 (m, aromatic, 5H), 9.4 (s, NHH, 2H), the latter signal disappeared on addition of deuterium oxide; uv (ethanol): λ max 260 nm (ε 16570), 334 (5000).

*Anal.* Calcd. for C<sub>13</sub>H<sub>13</sub>N<sub>7</sub>O: C, 55.12; H, 4.63; N, 34.61. Found: C, 55.22; H, 4.73; N, 34.56.

### 1,2,2-Triacetyl-1-(7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazine (**4**).

A mixture of **1** and acetic anhydride (10 ml) was refluxed for 5 hours. The reaction mixture was cooled and diluted with water whereby a yellow precipitate was formed. The product was crystallised from benzene to give **4** as a yellow powder in 73% yield, mp 140-142°; ir (potassium bromide): 1720 cm<sup>-1</sup> (C=O); pmr (deuteriochloroform): δ 2.2 (s, N<sup>-</sup>COCH<sub>3</sub>, 3H), 2.7 (s, N<sup>2</sup> (COCH<sub>3</sub>)<sub>2</sub>, 6H), 6.9-7.8 (m, aromatic, 5H); uv (ethanol): λ max 268 nm (ε 27000), 316 (7000), 360 sh (600).

*Anal.* Calcd. for C<sub>17</sub>H<sub>17</sub>N<sub>7</sub>O<sub>3</sub>: C, 55.58; H, 4.66; N, 26.69. Found: C, 55.59; H, 4.59; N, 27.00.

### 3,8-Dimethyl-1-phenyl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazine (**5**).

A mixture of **3** (0.5 g), dry xylene (5 ml) and phosphoryl chloride (1 ml) was refluxed for 8 hours. The cooled reaction mixture was diluted with petroleum ether (bp 60-80°) and the supernatant liquid decanted. The residue was dissolved in water, ammonium hydroxide added and the precipitate was filtered off. The solid obtained was crystallised from ethanol to give **5** as red crystals in 79% yield, mp 257-259°; ir (potassium bromide): no characteristic bands for NH and CO groups; pmr (trifluoroacetic acid): δ 2.3 (s, CH<sub>3</sub>, 3H), 2.4 (CH<sub>3</sub>-C=N, 3H), 6.7-7.4 (m, aromatic, 5H); uv (dioxane): λ max 268 nm (ε 42500), 498 (1600).

*Anal.* Calcd. for C<sub>13</sub>H<sub>11</sub>N<sub>7</sub>: C, 58.86; H, 4.18; N, 36.96. Found: C, 59.01; H, 4.16; N, 36.88.

### 2-Benzoyl-1-(7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazine (**6**).

A mixture of **1** (0.3 g) and benzoyl chloride (5 ml) was heated on a water bath for 5 hours. The solid product thus obtained was filtered, washed with benzene and crystallised from ethanol to give **6** as yellow needles in 93% yield, mp 239-241°; ir (potassium bromide): 1650 cm<sup>-1</sup> (C=O), 3200 (NH).

*Anal.* Calcd. for C<sub>18</sub>H<sub>15</sub>N<sub>7</sub>O: C, 62.60; H, 4.38; N, 28.39. Found: C, 62.55; H, 4.44; N, 28.43.

### 3-Methyl-1,8-diphenyl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazine (**7a**). Method A.

A mixture of **6** (0.5 g), dry xylene and phosphoryl chloride was refluxed for 8 hours. The usual working up procedure gave **7** as red crystals in 84% yield, mp 258-260°; ir (potassium bromide): no absorption bands for NH and CO groups; pmr (trifluoroacetic acid): δ 2.3 (s, CH<sub>3</sub>, 3H), 6.6-6.7 (m, 1-Ar-H, 8-Ar-H (2',3',5',6'), 9H), 9.0 (s, 8-Ar-H (4'), 1H); uv (dioxane): λ max 285 nm (ε 56000), 516 (2200).

*Anal.* Calcd. for C<sub>18</sub>H<sub>13</sub>N<sub>7</sub>: C, 66.05; H, 4.00; N, 29.95. Found: C, 65.91; H, 3.87; N, 29.66.

### Method B.

Compound **6** (1 g) was heated at 240-245° for 15 minutes on a sand bath. The solid mass was extracted with benzene-petroleum ether (bp 60-80°) mixture and the extract was concentrated and cooled whereby red crystals were separated. The product was collected and recrystallized from benzene to give **7a** as red crystals in 32% yield, mp 259-260°.

### Method C.

Compound **1** (1 g) was refluxed with benzoyl chloride (10 ml) for 5 hours. Excess benzoyl chloride was distilled off under reduced pressure and the residue washed with hot petroleum ether (bp 60-80°). The product was collected and crystallised from benzene to give **7a** as red crystals in 37% yield, mp 258-260°.

### Method D.

Benzaldehyde (7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazone (**8a**) (0.5 g) and thionyl chloride were heated on a water bath for 3 hours. Excess thionyl chloride was removed by distillation and the residue washed with hot petroleum ether (bp 60-80°). The product was crystallised from benzene to give **7a** as red crystals in 68% yield mp 258-260°. The reaction product was found to be identical with those formed by the above methods in all aspects.

### 3-Methyl-1-phenyl-8-aryl-1*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazines **7b-e**. General Procedure.

A mixture of **8b-c** [11] (0.3 g) and thionyl chloride (10 ml) was heated on a water bath for 3 hours. Excess thionyl chloride was removed by distillation. The residue was triturated with petroleum ether (bp 60-80°) and the products were crystallised from the proper solvent to give **7b-e**

(Table I).

6-Methyl-8-phenyl-8*H*-pyrazolo[3,4-*e*]tetrazolo[5,1-*c*][1,2,4]triazine (**9**).

To compound **1** (0.25 g, 0.001 mole) in concentrated phosphoric acid (2 ml), sodium nitrite solution (1.5 ml, 0.005 mole) was added at 0° during 15 minutes with stirring. The mixture was stirred for a further 2 hours and the precipitate was filtered off and dried. The solid product was crystallised from methanol to give **9** as orange crystals in 96% yield, mp 138-140°; ir (potassium bromide): 1215 cm<sup>-1</sup> (tetrazole ring) [10], no absorption band at 2120-2150 characteristic for azido group; pmr (DMSO-*d*<sub>6</sub>): δ 2.6 (s, CH<sub>3</sub>, 3H), 6.9-7.8 (m, aromatic, 5H); uv (ethanol): λ max 258 nm (ε 52500), 444 (1500).

*Anal.* Calcd. for C<sub>11</sub>H<sub>8</sub>N<sub>6</sub>: C, 52.38; H, 3.20; N, 44.42. Found: C, 52.51; H, 3.37; N, 44.61.

1,7-Dihydro-3-methyl-1-phenyl-8*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazine-8-thione (**10**).

A mixture of **1** (0.3 g), methanol (20 ml), potassium hydroxide (0.1 g) and carbon disulphide (1 ml) was refluxed for 4 hours. The reaction mixture was filtered, concentrated and neutralised with acetic acid whereby a dark green material was precipitated. The product was crystallised from xylene to give **10** in 57% yield, mp > 300°; ir (potassium bromide): 1275 cm<sup>-1</sup> (C=S), 3100 (NH); uv (ethanol): λ max 248 nm (ε 11750), 286 (20500), 404 (600), 560 (475).

*Anal.* Calcd. for C<sub>12</sub>H<sub>8</sub>N<sub>4</sub>S: C, 50.87; H, 3.20; N, 34.61; S, 11.32. Found: C, 50.69; H, 3.11; N, 34.65; S, 11.39.

2-Ethoxycarbonyl-1-(7-methyl-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazin-3-yl)hydrazine (**11**).

A mixture of **1** (0.3 g) and ethyl chloroformate (0.22 ml) was heated in pyridine (20 ml) on a water bath for 8 hours. The reaction mixture was cooled and acidified with dilute acetic acid whereby a yellow precipitate was formed. The product was crystallised from ethanol to give **11** as yellow needles in 62% yield, mp 204-205°; ir (potassium bromide): 3320, 3220 cm<sup>-1</sup> (NH), 1750 (C=O); uv (ethanol): λ max 260 nm (ε 17250), 334 (5312).

*Anal.* Calcd. for C<sub>14</sub>H<sub>15</sub>N<sub>7</sub>O<sub>2</sub>: C, 53.67; H, 4.83; N, 31.29. Found: C, 53.76; H, 4.63; N, 31.25.

1,7-Dihydro-3-methyl-1-phenyl-8*H*-pyrazolo[3,4-*e*][1,2,4]triazolo[3,4-*c*][1,2,4]triazin-8-one (**12**).

Compound **11** (0.3 g) was heated on a sand bath at 205-210° for 10 minutes. The resulting material was crystallised from benzene to give **12** as violet product in 78% yield, mp 268-270°; ir (potassium bromide): 3180 cm<sup>-1</sup> (NH), 1710 (C=O); pmr (trifluoroacetic acid): δ 2.05 (s, CH<sub>3</sub>, 3H), 6.5-7.2 (m, aromatic, 5H), 9.1 (s, NH, 1H); uv (ethanol): λ max 226 nm (ε 8750), 264 (17500), 325 sh (450), 532 (625).

*Anal.* Calcd. for C<sub>12</sub>H<sub>9</sub>N<sub>7</sub>O: C, 53.93; H, 3.39; N, 36.69. Found: C, 53.80; H, 3.56; N, 36.51.

7-Methyl-3-(3-methyl-2-pyrazolin-1-yl-5-one)-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazine (**13**).

Compound **1** (0.3 g), ethyl acetoacetate (0.16 ml) and potassium hydroxide (0.1 g) in ethanol (20 ml) were refluxed for 3 hours. The reaction mixture was cooled and neutralized with acetic acid whereby a yellow material was precipitated. The product was crystallised from ethanol to give **13** as yellow crystals in 77% yield, mp 254-256°; ir (potassium bromide): 1660 cm<sup>-1</sup> (C=O).

*Anal.* Calcd. for C<sub>15</sub>H<sub>13</sub>N<sub>7</sub>O: C, 58.63; H, 4.26; N, 31.90. Found: C, 58.71; H, 4.36; N, 31.73.

7-Methyl-3-(3,5-dimethylpyrazol-1-yl)-5-phenyl-5*H*-pyrazolo[3,4-*e*]-1,2,4-triazine (**14**).

The above method was employed using acetylacetone instead of ethyl acetoacetate. The product was crystallised from ethanol to give **14** as greenish yellow powder in 53% yield, mp 136-138°; ir (potassium bromide): no absorption bands due to NH or NH<sub>2</sub> groups.

*Anal.* Calcd. for C<sub>16</sub>H<sub>15</sub>N<sub>7</sub>: C, 62.94; H, 4.95; N, 32.11. Found: C, 62.91; H, 5.14; N, 32.09.

## REFERENCES AND NOTES

- [1] To whom all correspondence should be addressed.
- [2] A. Monge Vega, I. Aldama, M. M. Rabbani and E. Fernandez-Alvarez, *J. Heterocyclic Chem.*, **17**, 77 (1980).
- [3] A. Messmer, Gy. Hajo's, P. Benko and L. Pallos, *J. Heterocyclic Chem.*, **10**, 575 (1973).
- [4] K. T. Potts and S. Husain, *J. Org. Chem.*, **36**, 10 (1971).
- [5] K. T. Potts and R. M. Huseby, *ibid.*, **31**, 3528 (1966).
- [6] J. Kobe, R. K. Robinson and D. E. O'Brien, *J. Heterocyclic Chem.*, **11**, 199 (1974).
- [7] J. Kobe, D. E. O'Brien, R. K. Robinson and T. Novenson, *J. Heterocyclic Chem.*, **11**, 911 (1974).
- [8] J. H. Bellary and V. V. Badiger, *Indian J. Chem.*, **20B**, 654 (1981).
- [9] N. A. Shams, A. M. Kaddah and A. H. Moustafa, *ibid.*, **21B**, 317 (1982).
- [10] K. C. Joshi and P. Chand, *J. Heterocyclic Chem.*, **17**, 1783 (1980).
- [11] M. S. K. Youssef, F. M. Atta, Kh. M. Hassan and M. S. Abbady, *J. Heterocyclic Chem.*, **21**, 923 (1984).
- [12] A. Messmer, G. Majo's, J. Tama and A. Neszmelyi, *J. Org. Chem.*, **44**, 1823 (1979).
- [13] M. Tisler, *Synthesis*, 123 (1973).